

## THEORY

1. Temperature is a state variable indicating whether or not an object is in thermal equilibrium with its environment. Temperature can be measured in different ways, by choosing a property of the thermometer (e.g. volume, resistance, radiation, etc.) which is a one-to-one function of temperature and easy to measure. To assign a numerical value to the temperature we have to scale the measured value. Choosing a working material and a property, with an arbitrary scale we get an empirical temperature. The thermodynamic temperature scale is based on the second law of thermodynamics (the efficiency of the Carnot engine).

The properties of a good thermometer are:

- small heat capacity (it does not change the temperature to be measured) ,
- small inertia (it reaches the thermal equilibrium fast),
- good reproducibility.

Thermometers can be grouped in the following way:

1.) The thermometer and the object with unknown temperature are not in direct connection. These are pyrometers for measuring high temperatures (above  $\sim 500^{\circ}\text{C}$ ) based on the thermal radiation.

2.) The thermometer and the object are in direct connection. These are mechanic and electric thermometers:

### 2.A.) Mechanic thermometers:

- Metal rod thermometer. The linear thermal expansion is measured.
- Bimetallic thermometer. Two strips of metals with different coefficients of linear expansion (e.g. brass and iron) are riveted together. When heated, the strip bends, when cooled, the coil opens up.
- Liquid-in-glass thermometers. The liquid may be ethanol, mercury or pentane. The expansion of the liquid in the reservoir can be read on the linear scale of the capillary joining with the reservoir.
- Mercury (liquid)-in-steel thermometer. Here the liquid is expanding in a flexible spiral tube the shape of which depends on the temperature.
- Vapour pressure thermometer. A suitable liquid is placed in a bulb and connected to a pressure gauge which is used graduated for temperature.
- Gas thermometer. The pressure in constant volume conditions (or volume in constant pressure conditions) is proportional to the thermodynamic temperature (supposing that the gas is ideal).

### 2.B.) Electric thermometers:

#### 2.B.1) Thermoelectric thermometer, thermocouple.

When two different metals join, an electric potential difference will occur at the junction point. This contact potential depends on the two materials and on the temperature. In a closed circuit the sum of these contact potentials equals to zero if the temperature of all joining points is equal. But if the joining points have different temperatures then a thermoelectromotive force is generated this way.

#### 2.B.2) Resistance thermometers.

Resistance thermometers are made of

- a) **metals**: The metals used for resistance thermometers are mainly Ni or Pt. Their inertia is relatively great.
- b) **semiconductors (called thermistors)**: In case of semiconductors the resistance is a nonlinear function of the temperature. Thermistors have greater sensitivity and smaller inertia than metal resistance thermometers have.

## 2. METAL RESISTANCE THERMOMETERS

The **resistance of metals** is a linear function of the temperature:

$$R(T) = R_0 ( 1 + \beta(T-T_0) ) , \quad \text{where}$$

$R_0$  is the resistance at the reference temperature  $T_0$ ,

$\beta$  is the temperature coefficient. [1/C°] or [1/K]

In case of metals  $\beta$  can be considered constant for a certain temperature range.

This is the sensitivity of metal resistance thermometer as  $\beta = \frac{\frac{\Delta R}{R_0}}{\Delta T}$ .

In this measurement a Pt resistance thermometer will be used.

## 3. THE INERTIA OF THERMOMETERS

When the temperature of the thermometer's environment suddenly changes then the thermometer approaches this new value gradually.

The process can be described by **Newton's equation**:

$$\Delta T(t) = \Delta T_0 \cdot e^{-\frac{t}{\tau}} , \quad \text{where}$$

$\Delta T$  is the temperature **difference**

**between the thermometer and its environment** at the time instant  $t$ ,

$\Delta T_0$  is the initial temperature difference (at  $t_0 = 0$ ),

$\tau$  is the time constant of the thermometer [s].

Substituting  $t = \tau$  into Newton's equation gives  $\Delta T(\tau) = \frac{\Delta T_0}{e}$ , i.e. the **time constant** is the time elapsed while the temperature difference between the thermometer and its environment decreases in value to  $1/e$  ( $\approx 36.8\%$ ).

The time constant  $\tau$  depends on

- the heat capacity of the thermometer  $C$ ,
- the surface of the thermometer (through which the heat transport happens)  $A$ ,
- the heat transport coefficient between the thermometer and the environment  $\alpha$ :

$$\tau = \frac{C}{A \alpha} .$$

Instead of  $\tau$  sometimes  $t_{1/2}$  (the half-time) is used: at  $t = t_{1/2}$   $\Delta T(t_{1/2}) = \frac{\Delta T_0}{2}$ .

With this constant Newton's equation is written as  $\Delta T = \Delta T_0 \cdot 2^{-\frac{t}{t_{1/2}}}$ .

## MEASUREMENT

We shall measure temperature-fall and -raise curves with a Pt resistance thermometer. The resistance is measured with a digital multimeter.

First measure the temperature-fall curve:

- check that the resistance of the Pt resistance thermometer is constant in the thermostat,
- then as fast as possible put the thermometer into the jar of water with ice cubes, and
- record the resistance as a function of time, at the time instants indicated in the table.

Measure the temperature-raise curve in a similar way.

## EVALUATION

Use Excel for performing all calculations and for plotting the graphs.

1. Calculate the temperature at every time instant from the resistance using the formula

$$R(T) = R_0 ( 1 + \beta (T - T_0) ) , \text{ where}$$

$T_0 = 0 \text{ }^\circ\text{C}$  (the temperature of the water with ice),

$R_0$  is the resistance measured in the water with ice,

$\beta = 0.00386 \text{ } 1/\text{C}^\circ$  is the value of the temperature coefficient of Pt.

The temperature of the thermostat  $T_{\text{therm}}$  is calculated from the resistance  $R_{\text{therm}}$ .

2. Calculate  $\Delta T$  for the temperature fall and raise curves:  $\Delta T_{\text{fall}} = T - T_0$ , and  $\Delta T_{\text{raise}} = T_{\text{therm}} - T$ .

3. Plot the two  $\Delta T$  vs.  $t$  graphs.

4. Determine the two time constants  $\tau_{\text{fall}}$  and  $\tau_{\text{raise}}$  in the following ways:

a) Fit exponential trendline to the  $\Delta T$  vs.  $t$  graphs. Neglect those points that noticeably do not fit to the trendline. The value of the time constant can be calculated from the exponent ( as  $\Delta T = \Delta T_0 \cdot e^{-\frac{t}{\tau}}$  ).

b) Linearize the exponential equation  $\Delta T = \Delta T_0 \cdot e^{-\frac{t}{\tau}}$ . Calculate the slope using the least squares method, then calculate the value of  $\tau$  from the slope.

5. Summarize the results in your report.